## Final Examination May 2, 2017

## **Problem 1. (50%)**

Consider a continuous stirred-tank reactor operated under isothermal conditions in which the following three reactions take place:

$$A + B \xrightarrow{k_1} C \qquad A + C \xrightarrow{k_2} D \qquad B + C \xrightarrow{k_3} E$$
(1)

The rates of the three irreversible reactions are given by an elementary mechanism,

$$r_1 = k_1[A][B]$$
  $r_2 = k_2[A][C]$   $r_3 = k_3[B][C]$  (2)

The material balances for the reactor are given by (accumulation = in – out +/- generation/consumption):

$$\frac{d[A]}{dt} = F[A]_{in} - F[A] - r_1 - r_2$$
(3.A)

$$\frac{d[B]}{dt} = F[B]_{in} - F[B] - r_1 - r_3$$
(3.B)

$$\frac{d[C]}{dt} = F[C]_{in} - F[C] + r_1 - r_2 - r_3$$
(3.C)

$$\frac{d[D]}{dt} = F[D]_{in} - F[D] + r_2$$
(3.D)

$$\frac{d[E]}{dt} = F[E]_{in} - F[E] + r_3$$
(3.E)

Consider a specific case where the volumetric flowrate per unit volume F = 0.1 1/s and the inlet concentrations  $[A]_{in} = [B]_{in} = 1.0$  mol/liter &  $[C]_{in} = [D]_{in} = [E]_{in} = 0.0$  mol/liter. Assume the reactor is initially empty, [A] = [B] = [C] = [D] = [E] = 0.0 mol/liter at t = 0.

An experiment is performed in which the concentrations of the five species are measured as a function of time. This data is recorded in the file located at <u>http://www.utkstair.org/clausius/docs/mse510/data/mse510\_xm02\_p01\_s17.txt</u>. The data file contains six columns corresponding to time, and the concentrations of A, B, C, D and E.

Using this information, determine the optimal values of the three reaction rate constants,  $k_1$ ,  $k_2$  and  $k_3$ . The magnitude of these rate constants should all be on the order of one. Also report the value of the objective function.

## Problem 2. (50%)

The one-dimensional heat equation can describe heat transfer in a rutile  $TiO_2$  with both heat conduction and radiative heat loss.

$$\frac{\partial T}{\partial t} = \frac{k}{\rho C_p} \frac{d^2 T}{dz^2} - \frac{\varepsilon \sigma S}{\rho C_p} \left(T^4 - T_s^4\right)$$

where the following variables [with units] are given as temperature in the material *T* [K] surrounding temperature  $T_s = 800$  [K] axial position along material *z* [m] thermal conductivity k = 9.0 [J/K/m/s] mass density  $\rho = 4250$  [kg/m<sup>3</sup>] heat capacity  $C_p = 711.3$  [J/kg/K] Stefan–Boltzmann constant  $\sigma = 5.670373x10^{-8}$  [J/s/m<sup>2</sup>/K<sup>4</sup>] gray body permittivity  $\varepsilon = 0.79$ surface area to volume ratio S = 200 [m<sup>-1</sup>] (for a cylindrical rod of diameter 0.05 m)

A rutile TiO<sub>2</sub> cylindrical rod of diameter 0.05 m and length 0.4 m is initially at T(z,t=0) = 1000 K. One end of the rod is maintained at T(z=0,t) = 1000 K. The other end of the rod is maintained at T(z=L,t) = 1500 K.

(a) Sketch the transient behavior.

(b) Find the approximate steady-state temperature in the material at z=0.2 m.